

Calculate the rotational absorption spectra of the ^{13}CO molecule

1. Determine the ^{13}CO moment of inertia
 - a. Find the ^{13}CO bond length (assume it is the same length as the CO bond)
 - b. Calculate the center of mass of the ^{13}CO molecule
 - c. Calculate the moment of inertia of the ^{13}CO molecule
2. Use the simple dumbbell rotational spectrum model to determine the first 26 rotational energy levels of ^{13}CO
3. Use these to determine the 25 lowest rotational line spectra in both Hz and cm^{-1} .
Based on the JPL molecular spectroscopy catalog (spec.jpl.nasa.gov), calculate how close these frequencies are to the actual rotational line frequencies in a fractional sense:
 $(f_{\text{model}} - f_{\text{actual}}) / f_{\text{actual}}$.